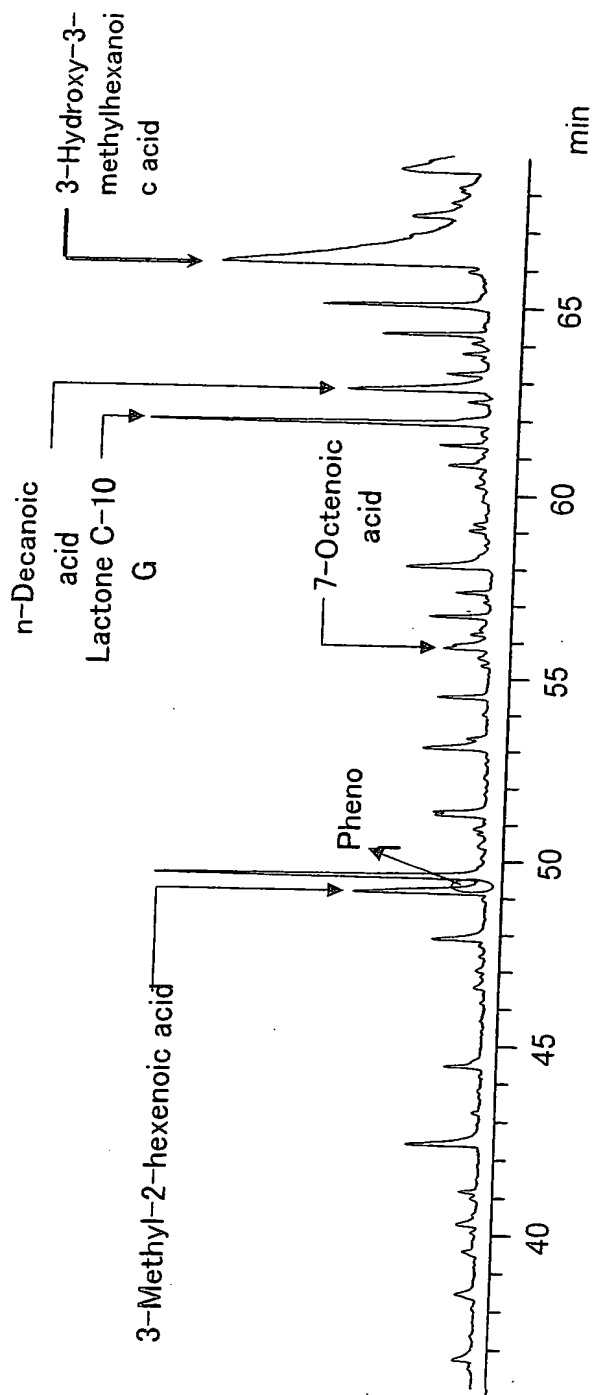
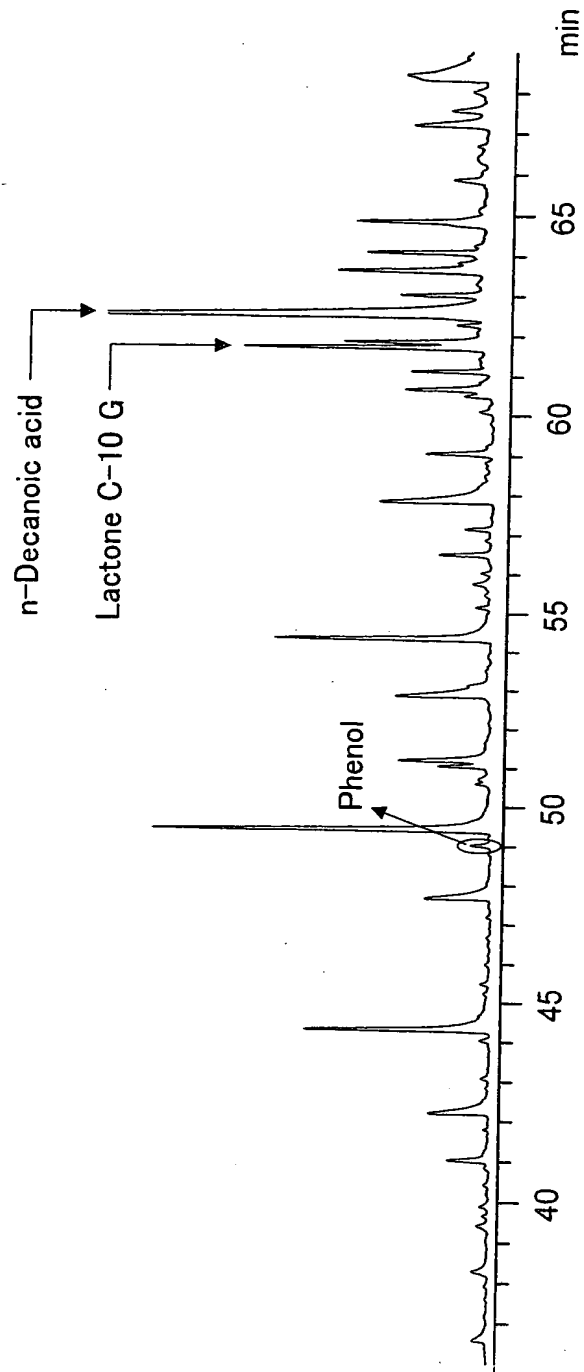


FIG. 1



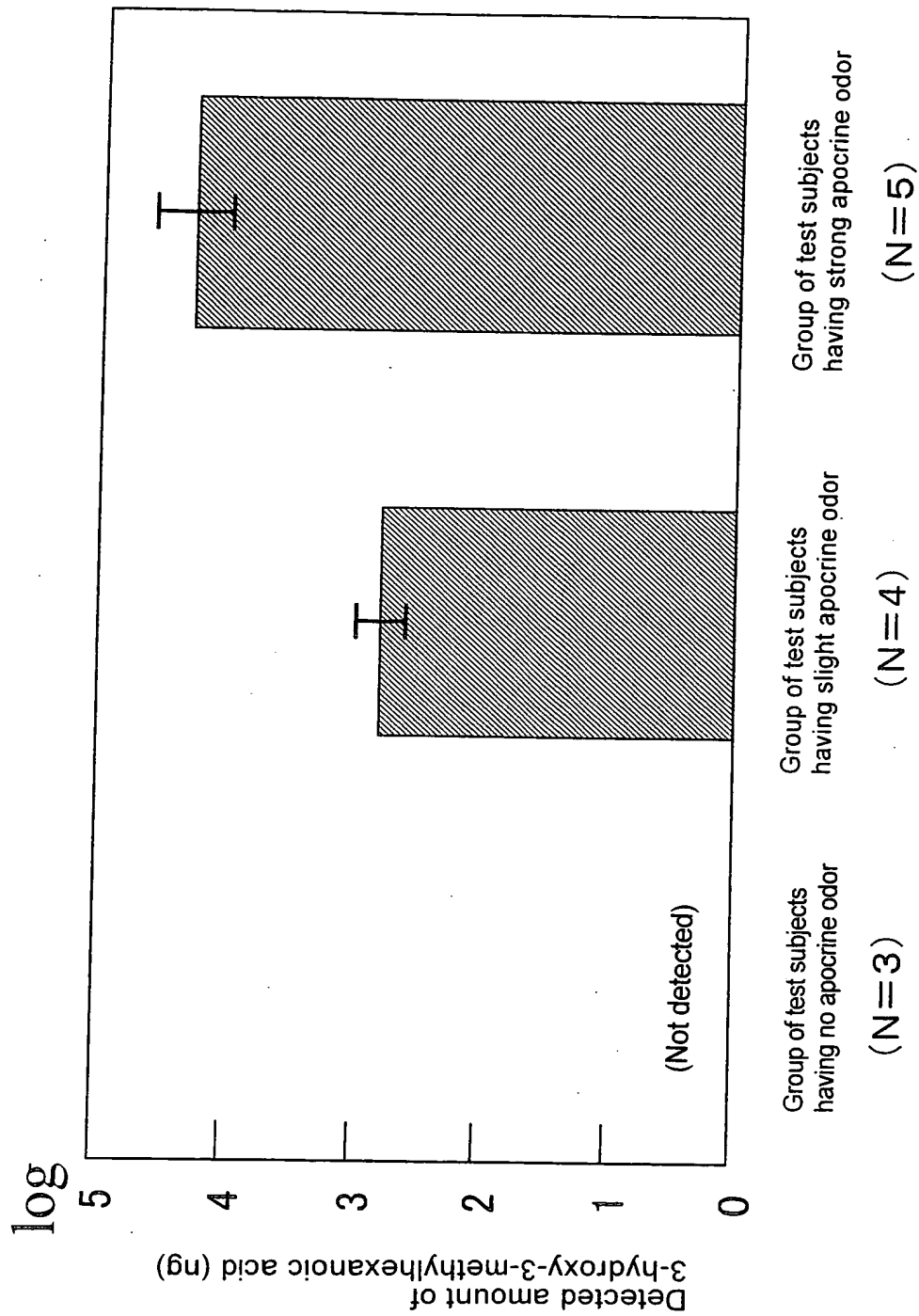
Total ion chromatogram of an acid extract of perspiration  
of a person having an apocrine odor

FIG. 2



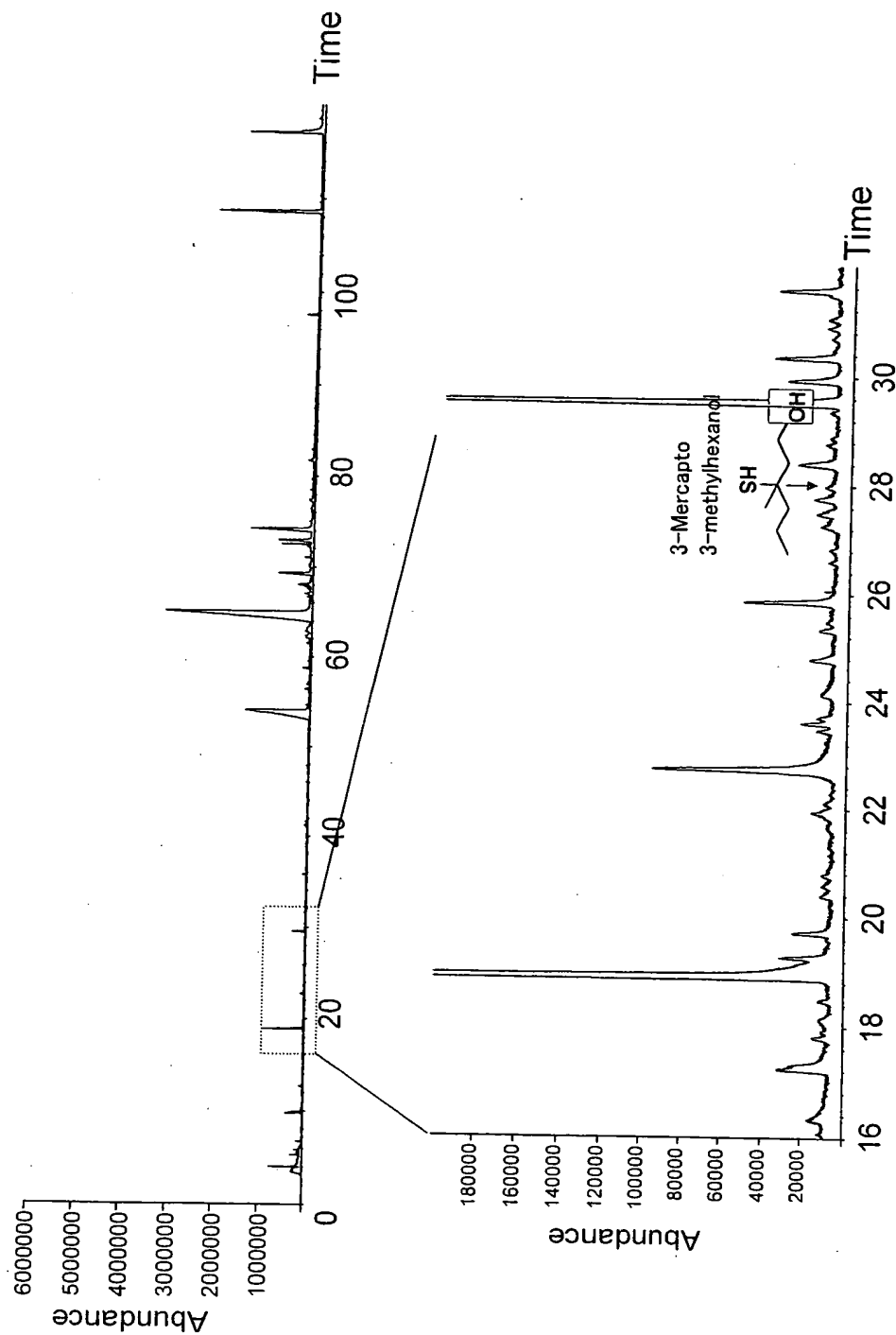
Total ion chromatogram of an acid extract of perspiration  
of a person having no apocrine odor

FIG. 3



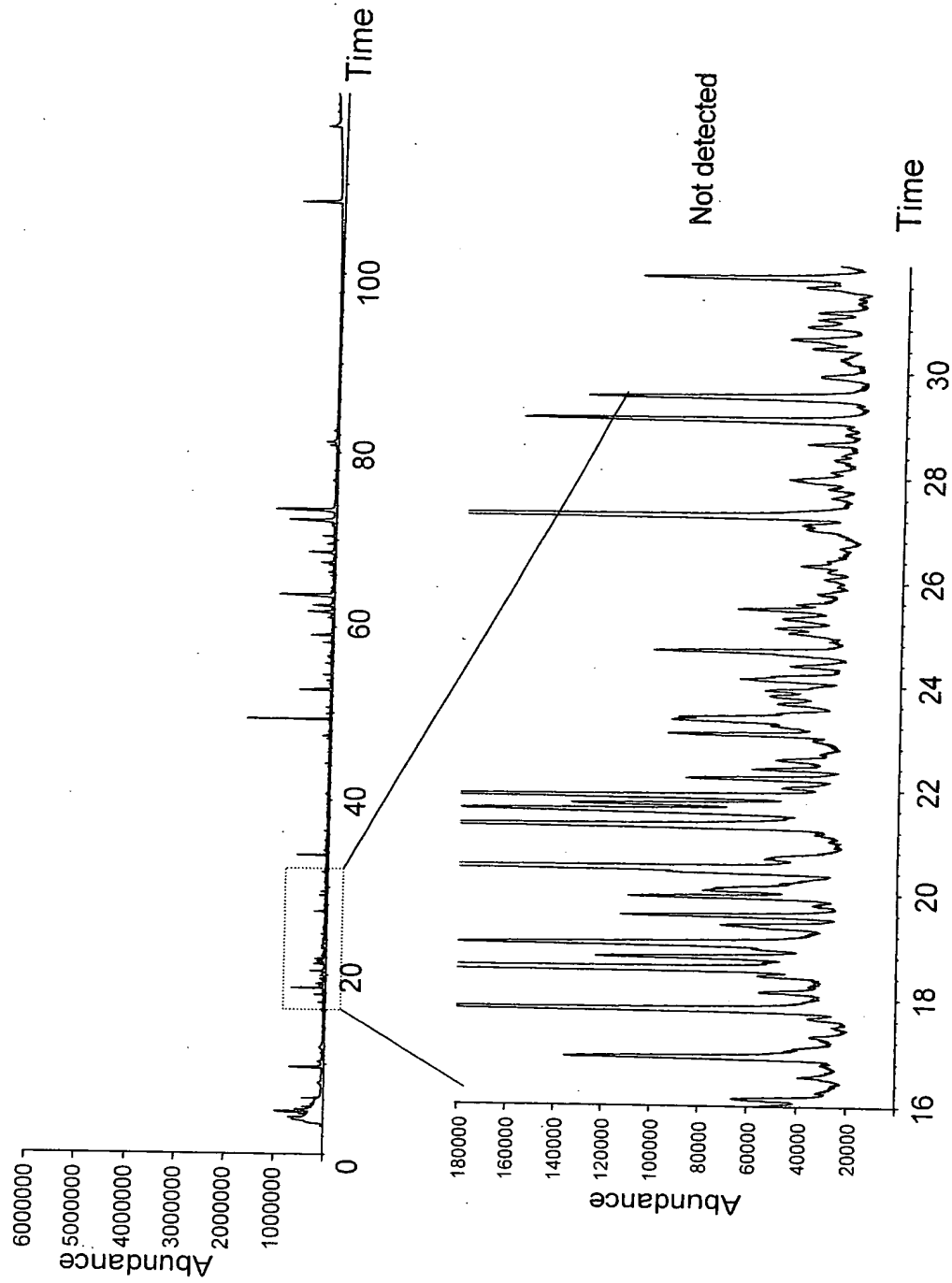
Relationship between 3-hydroxy-3-methylhexanoic acid  
and an apocrine odor

FIG. 4



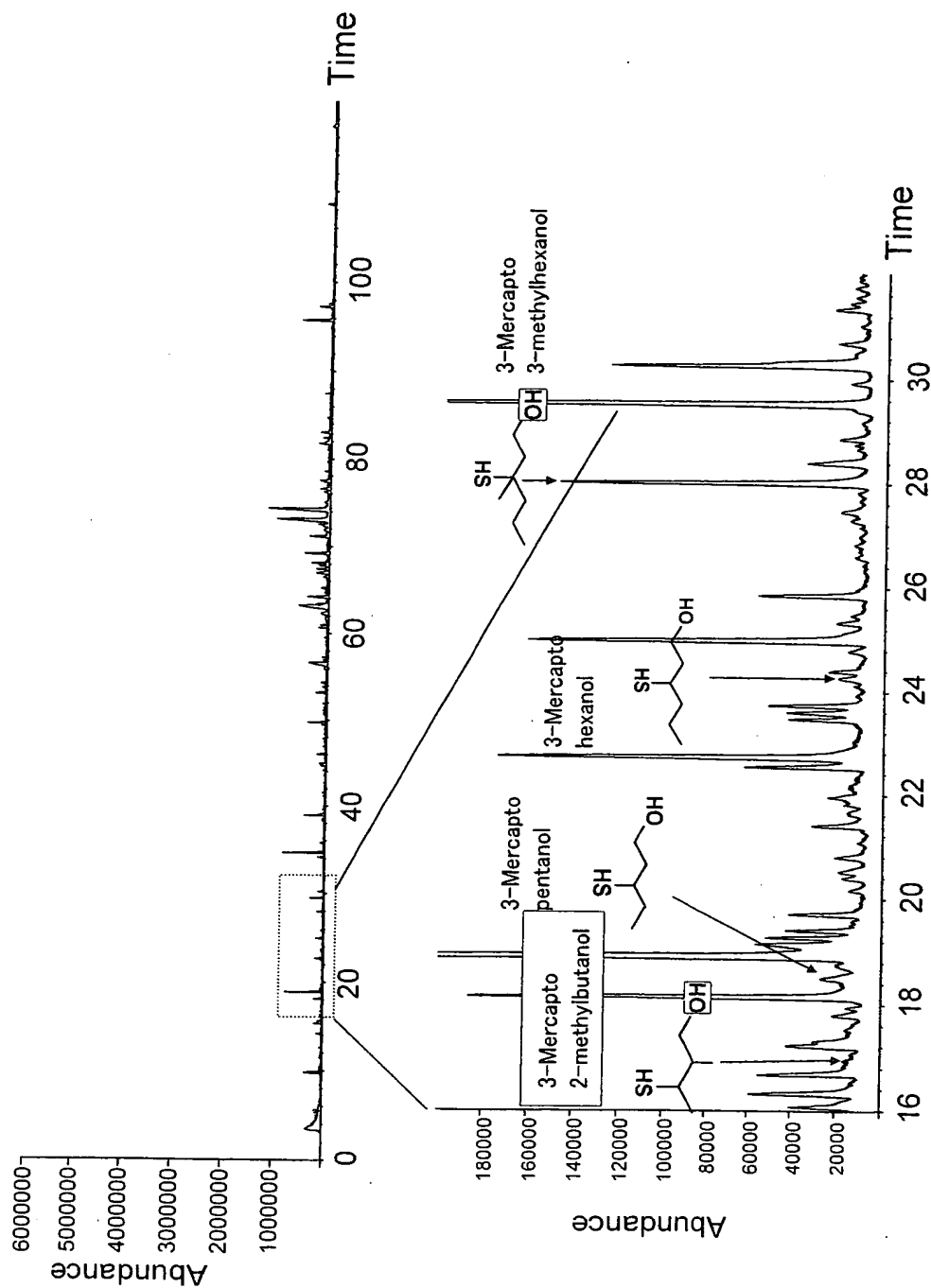
Total ion chromatogram of volatile components in perspiration  
of a person having an apocrine odor

FIG. 5



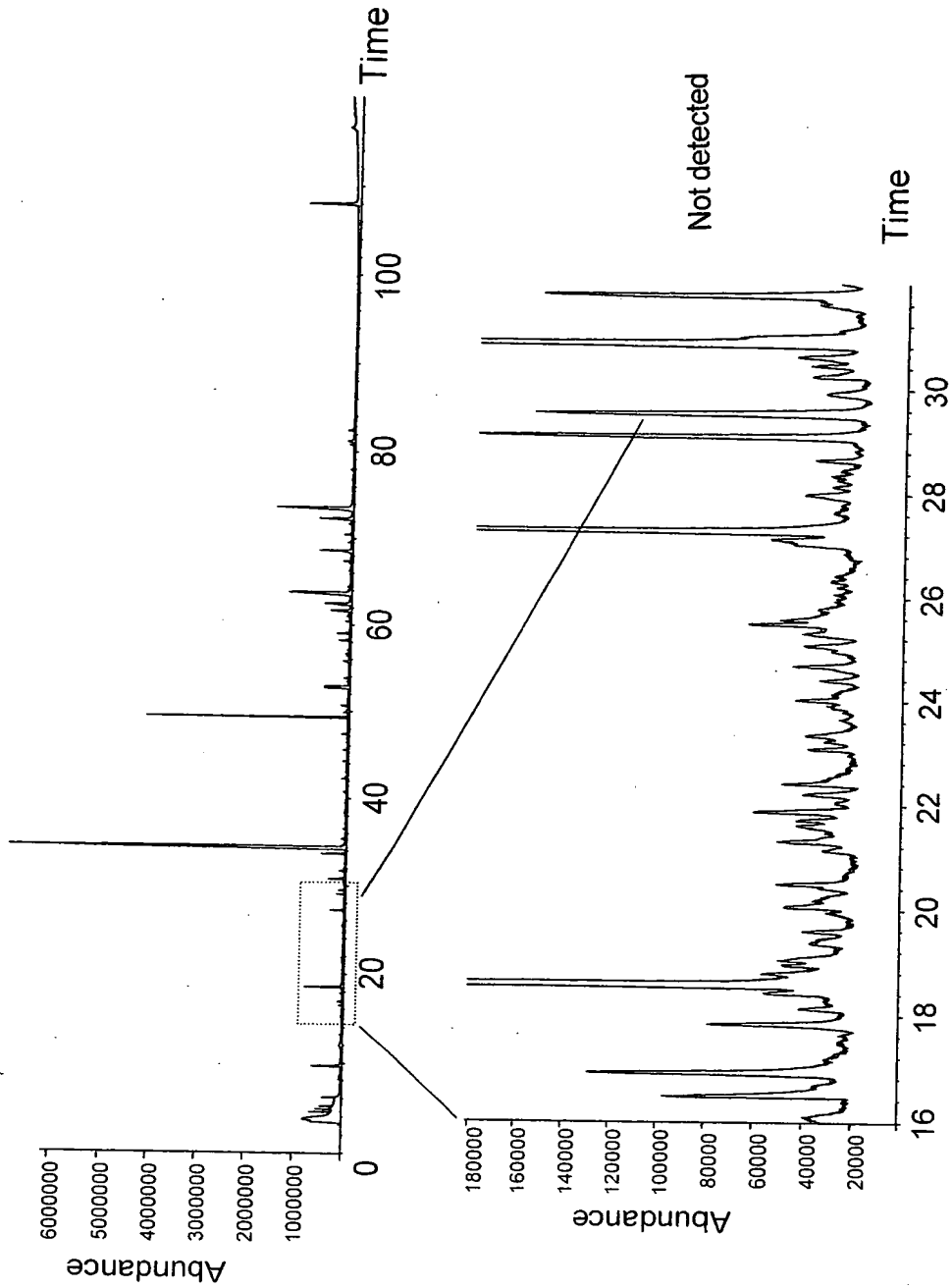
Total ion chromatogram of volatile components in perspiration  
of a person having no apocrine odor

FIG. 6



Total ion chromatogram of volatile components in incubated perspiration  
 of a person having an apocrine odor

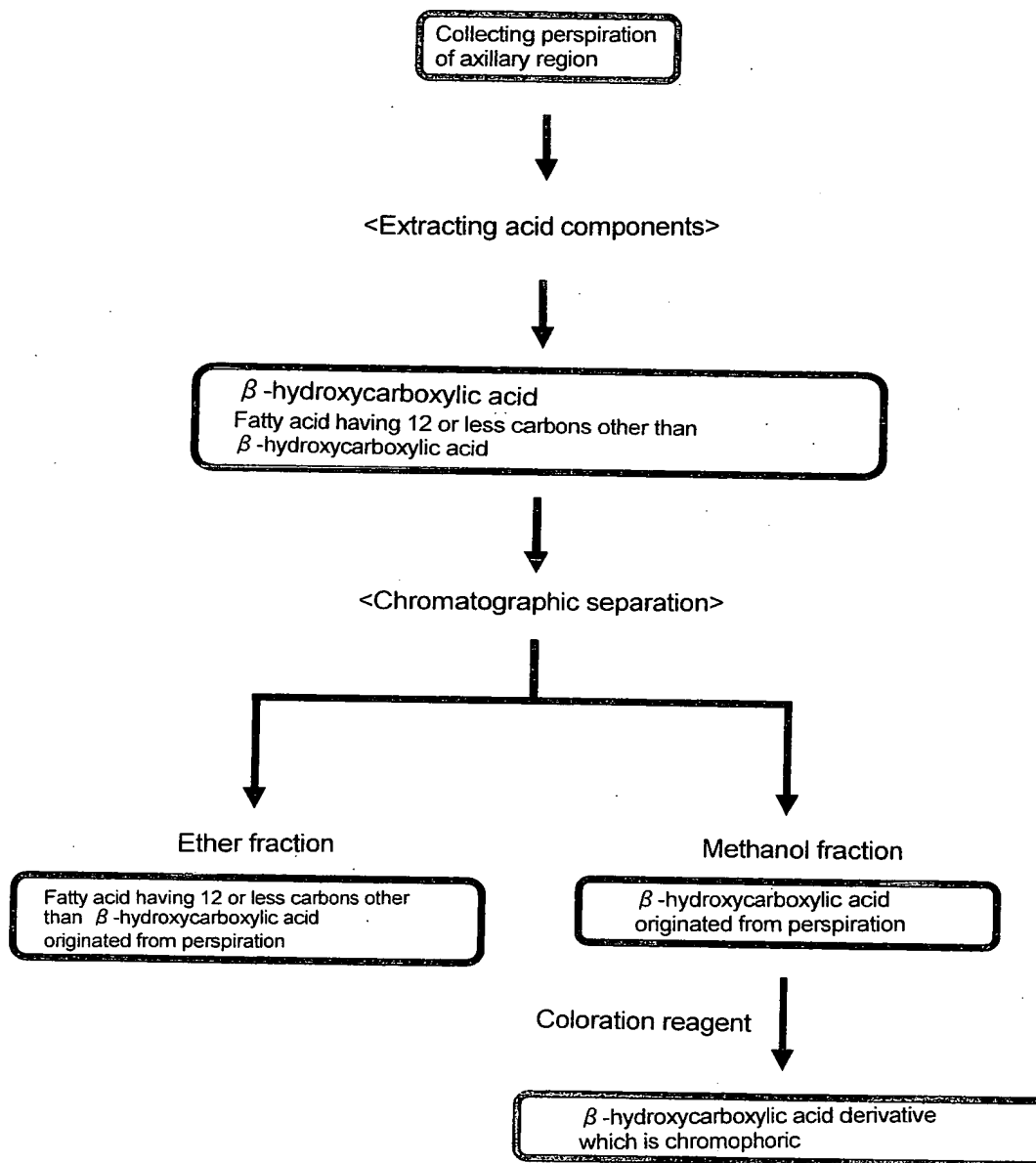
FIG. 7



Total ion chromatogram of volatile components in incubated perspiration  
of a person having no apocrine odor

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FIG. 8

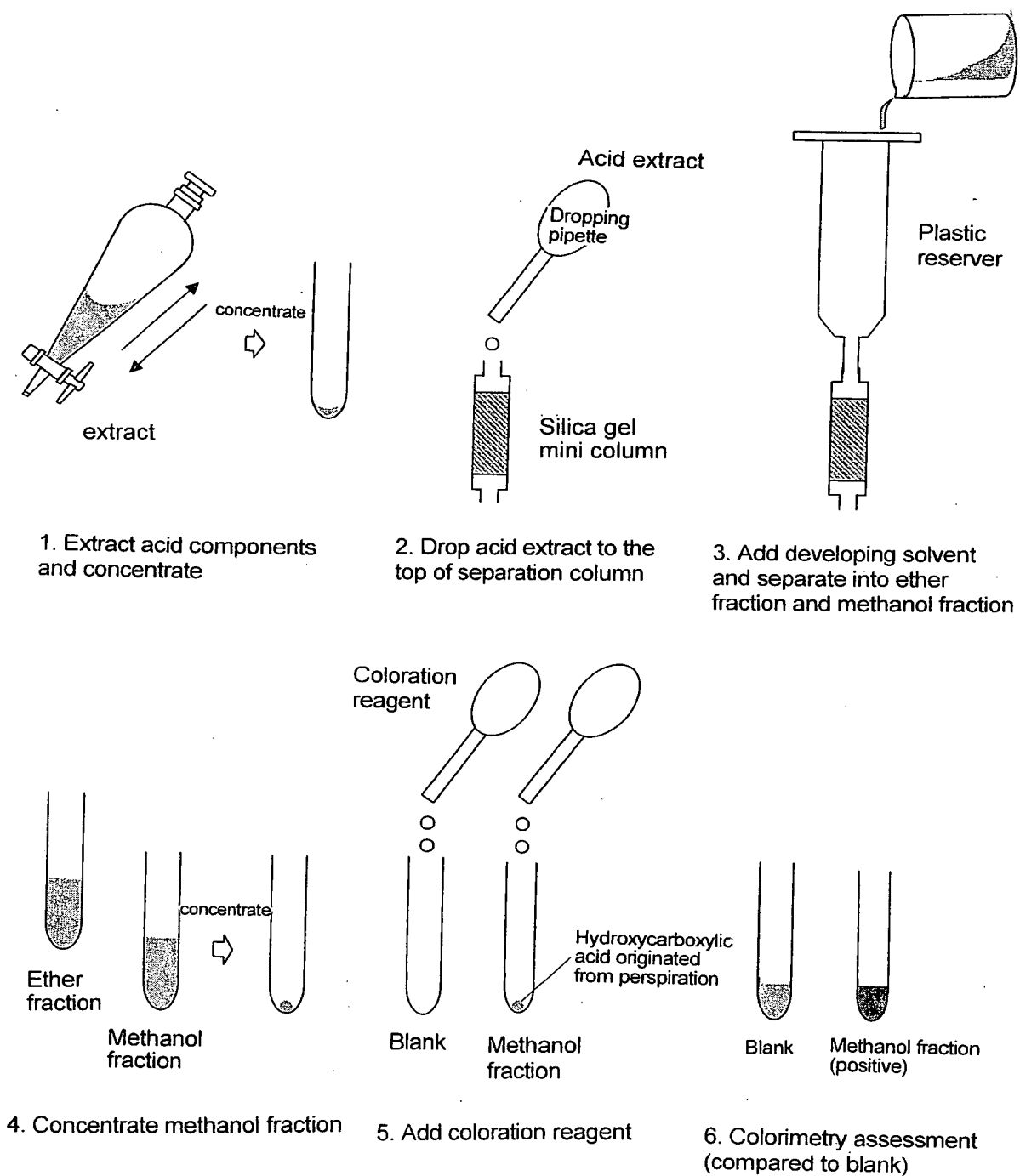


Quick and convenient method for examining body odor  
utilizing the coloration reaction



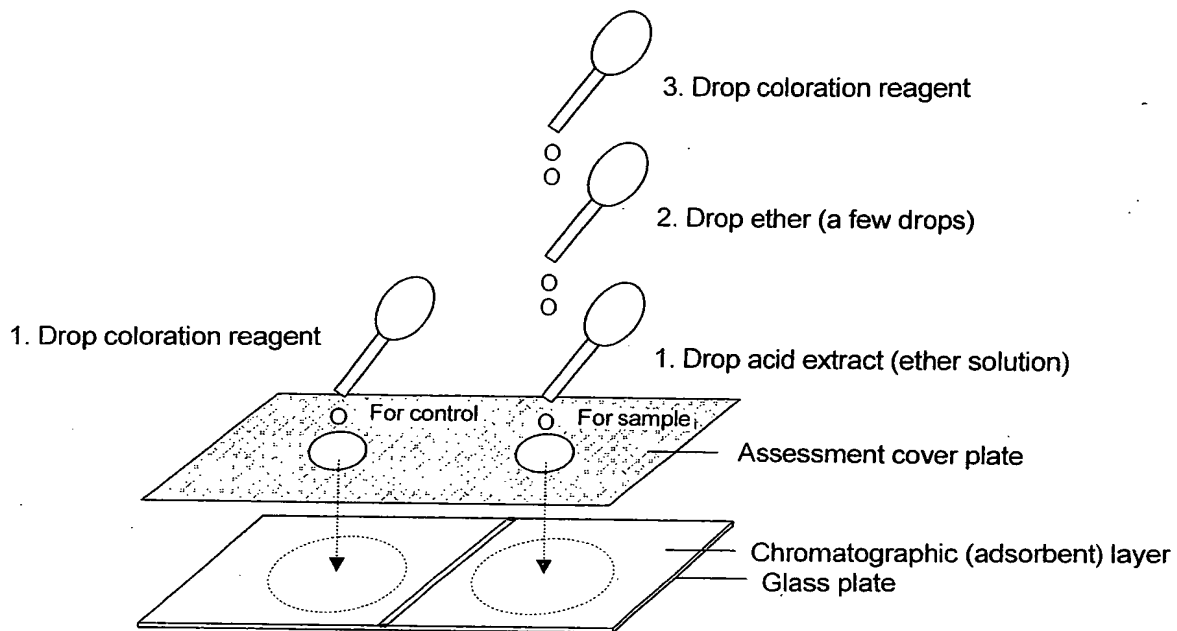
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FIG. 9

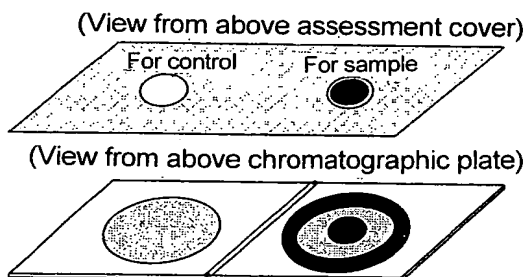


Example (1) of how kit for conveniently accessing body odor is used

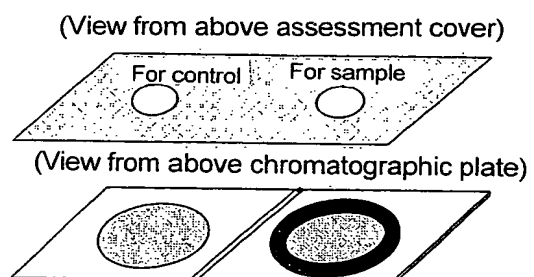
FIG. 10



<If positive>

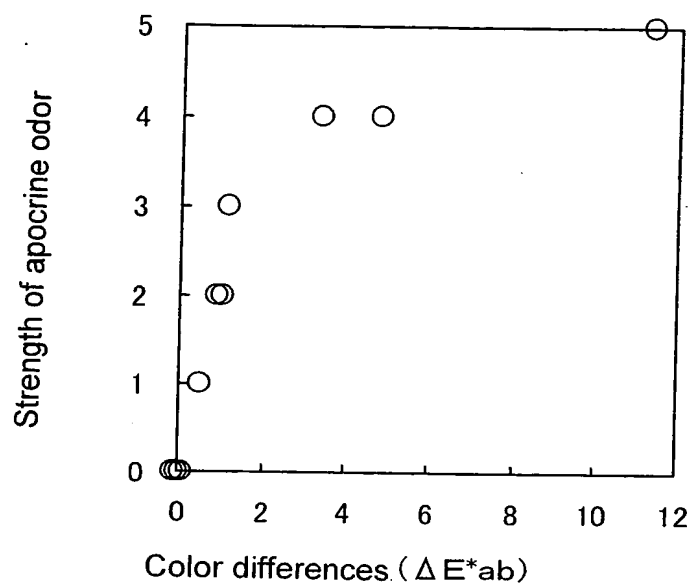


<If negative>



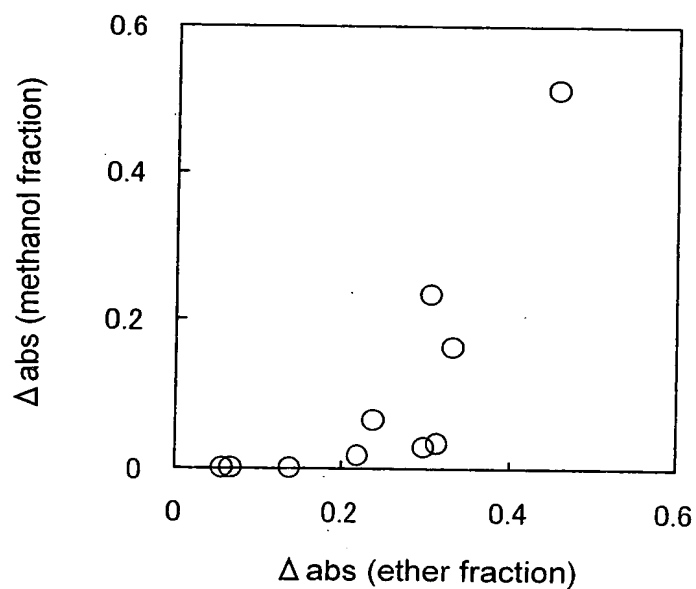
Example (2) of how kit for conveniently accessing body odor is used

FIG. 11



Relationship between strength of  
apocrine odor and color differences

FIG. 12



Absorbance of methanol  
fraction and ether fraction